



Summer Assignment Chemistry

Directions: Welcome to Chemistry. Below you will find some introductory assignments to prepare you for the first couple of chapters of the course and to lay a solid base for the remainder of the school year. Please contact Mr. Frageau if you have any questions at mfrageau@knoxschool.org.

This overall summer assignment is broken down into different partial assignments by number. It is recommended that you do 1 assignment per week. This is in order to give you some sort of time frame in order to help you not try to rush everything right before the school year begins. The entire document is due by September 6, 2022 through email to Mr. Frageau (mfrageau@knoxschool.org).

Academic Integrity: All work should be completed independently and without the assistance of unapproved resources. Any work violating academic integrity will be subject to a “0” and any additional consequences as outlined in the Knox Academic Integrity Policy attached to this assignment.

Due Date: Your work is due the first day of your Chemistry class. All late work will be subjected to a grade reduction or penalty as outlined in the course syllabus and copied below:

All major assignments not submitted on the due date will face a 10% deduction of max points per day for up to five (5) days and up to a 50% deduction.

Summer Assignments for AP Classes that are not submitted on time will result in the student being dropped from the course.

If you have any questions or concerns regarding this assignment, please contact the Dean of Academics, Mrs. Pergola, at dpergola@knoxschool.org .

Assignment 1.

1. Define the following terms **and** pick any 5 of the terms to write a sentence with. At the end of all your definitions write your 5 sentences and underline or highlight each of the terms as they are used.
 - a. Words to pick from: Chemistry, Mass, Volume, independent variable, dependent variable, control, theory, law, hypothesis, unit, density, scientific notation, dimensional analysis, accuracy, precision, error, matter, mole and significant figure.
2. What is the branch of chemistry that involves matter that does not contain carbon?
3. What branch of chemistry involves the energy and heat movement within a chemical reaction?
4. What branch of chemistry studies the composition or components of various substances?
5. What branch of chemistry studies matter and the processes within living organisms?

Assignment 2.

1. Go to the American Chemical Society homepage by following the website (<https://www.acs.org/content/acs/en.html>). Once there, scroll half way down the page and look for the area that says Molecule of the Week. Click on the yellow box that says "what molecule am I?" This will bring up the current molecule of the week information. You are to read the information about this substance and complete the following:
 - a. What is the date ? _____
 - b. What is the name of the substance? _____
 - c. What is the formula of this material (it is listed on the page)? _____
 - d. Write a paragraph about what you learned regarding this substance of the week.
 - e. Hand draw the picture / structure of the substance. (shown on the page)

Assignment 3.

1. Explain the process of the Scientific Method and give an example of when it might be used.
2. Explain the difference between a manipulated variable and a responding variable. After your response, write the terms and next to each one write down the vocabulary term from the list of week 1 which means the same thing.

Manipulated Variable = _____

Responding Variable = _____
3. ThoughtCo is an informative website with many different types of concepts. Visit the page (<https://www.thoughtco.com/reasons-to-study-chemistry-609210>) and describe 3 things that you learned about chemistry from this article.
 - a. .
 - b. .
 - c. .

Assignment 4.

1. Visit the SciWorthy website at (<https://sciworthy.com/o-the-drama/>) and read through the posted article. Upon completion of your reading, summarize the article to the best of your ability in a well developed paragraph.

Assignment 5.

1. What is the density of a material that has a volume of 24 mL and a mass of 30 grams?
2. What is the volume of liquid inside a container when the liquid has a mass of 81 grams and density of 2.4 g/ml ?
3. What is the mass of a gas that has a density of 0.51 g/ml and occupies a space of 11.4 mL?
4. Convert 2.8 hours to minutes. (show work setup)
5. Convert 1.93 kilograms to centigrams. (show work setup)
6. What does a prefix “ M “ represent when placed in front of “gram” ? (Mg or Mgram)
7. What does the prefix “ c “ represent when placed in front of “liter” ? (cL, cl or cliter)

Assignment 6.

1. Create a line graph for the information provided in the table and then answer questions relating to the data. (either by hand or digitally by using excel or google sheets)

Weekly June through July Rainfall in centimeters	
City 1	City 2
2.2	0.8
2.0	1.0
1.6	1.8
1.8	1.6
1.5	1.9
1.8	2.0
1.4	2.3
1.1	2.8

2. What is the independent variable?
3. What is the dependent variable?
4. What does it mean when the two lines merge?

Assignment 7.

1. A shopping mall wanted to determine whether the more expensive “Tough Stuff” floor wax was better than the cheaper “Steel Seal” floor wax at protecting its floor tiles against scratches. One liter of each

brand of floor wax was applied to test sections of the main hall of the mall. The test sections were all the same size and were covered with the same kind of tiles. After 3 weeks, the number of scratches in each of the test sections was counted to observe the wax's effectiveness.

Identify:

Independent variable: _____

Dependent variable: _____

Constants: _____

2. Mr. Smith wanted to see if the color of light shining on a plant had an effect on the number of leaves it had. He gathered 2 groups of the same species of plants, gave them the same amount of water, and did the test for the same amount of time. On one group of plants he used white light. On the second group he changed the light color to red. Identify:

Independent variable: _____

Dependent variable: _____

Constants: _____

Control group (What would Mr. Smith use as a control group?):

3. When running an experiment, you find that your data does not fit with what your original hypothesis was. What should be your next step? Explain.

Assignment 8.

- Which process is a physical change?
 - Water evaporating
 - A fire burning a wood log
 - A piece of paper turning color and crumpling due to sitting in the sun.
 - Light being released by a glow stick
- Which process is a chemical change?
 - Melting ice
 - Boiling water
 - Water breaking down into hydrogen and oxygen
 - Ice subliming into a gas
- How many significant digits are in the following number, 20.070 km?
 - 2
 - 3
 - 4
 - 5
- Which Kelvin temperature is equal to -24°C ?
 - 226 K
 - 249 K
 - 273 K
 - 297 K
- How many significant digits are in the number 0.00304 ?

- a. 5
- b. 4

- c. 3
- d. 2

6. Which of the following is NOT a SI base unit?

- a. Second
- b. Kilogram

- c. Degree celsius
- d. Meter

7. Which one of the following is the correct representation of 702.0 grams written in proper scientific notation?

- a. 7.02×10^3
- b. 70.20×10^1

- c. 7.020×10^2
- d. 70.20×10^2

8. When working with chemicals in the lab, which is something that you should NOT do?

- a. Read all labels on chemical bottles
- b. Pour unused materials back into the original bottle
- c. Use lots of water to wash skin that has been splashed with chemicals
- d. Take only as much as you need of shared chemicals

9. 3 dimensional volume is calculated by

- a. $D \times V \times mL$
- b. Mass * Volume
- c. Length * Width * Height
- d. Length * depth * mass

Assignment 9.

1. Complete the chart below by writing the elemental symbol, element name, and classification of element (metal, nonmetal, semimetal). Use a periodic table. If you do not have one, a recommended one can be found at ptable.com

Element #	Symbol	Name	Classification
1	H	Hydrogen	Nonmetal
2			
3			
4			
5			
6			

7			
8			
9			
10			
11			
12			
13			
14			
15			
16			
17			
18			
19			
20			
21			